Thiol-oxidant monochloramine mobilizes intracellular Ca²⁺ in parietal cells of rabbit gastric glands

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Walsh BM, Naik HB, Dubach JM, Beshire M, Wieland AM, Soybel DI. Thiol-oxidant monochloramine mobilizes intracellular Ca²⁺ in parietal cells of rabbit gastric glands. Am J Physiol Cell Physiol 292: C1687-C1697, 2007. First published February 7, 2007; doi:10.1152/ajpcell.00189.2006.—In Helicobacter pylori-induced gastritis, oxidants are generated through the interactions of bacteria in the lumen, activated granulocytes, and cells of the gastric mucosa. In this study we explored the ability of one such class of oxidants, represented by monochloramine (NH₂Cl), to serve as agonists of Ca²⁺ accumulation within the parietal cell of the gastric gland. Individual gastric glands isolated from rabbit mucosa were loaded with fluorescent reporters for Ca²⁺ in the cytoplasm (fura-2 AM) or intracellular stores (mag-fura-2 AM). Conditions were adjusted to screen out contributions from metal cations such as Zn²⁺, for which these reporters have affinity. Exposure to NH₂Cl (up to 200 µM) led to dose-dependent increases in intracellular Ca2+ concentration ([Ca²⁺]_i), in the range of 200–400 nM above baseline levels. These alterations were prevented by pretreatment with the oxidant scavenger vitamin C or a thiol-reducing agent, dithiothreitol (DTT), which shields intracellular thiol groups from oxidation by chlorinated oxidants. Introduction of vitamin C during ongoing exposure to NH2Cl arrested but did not reverse accumulation of Ca²⁺ in the cytoplasm. In contrast, introduction of DTT or N-acetylcysteine permitted arrest and partial reversal of the effects of NH₂Cl. Accumulation of Ca²⁺ in the cytoplasm induced by NH2Cl is due to release from intracellular stores, entry from the extracellular fluid, and impaired extrusion. Ca²⁺-handling proteins are susceptible to oxidation by chloramines, leading to sustained increases in [Ca2+]i. Under certain conditions, NH₂Cl may act not as an irritant but as an agent that activates intracellular signaling pathways. Anti-NH₂Cl strategies should take into account different effects of oxidant scavengers and thiol-reducing

calcium; Helicobacter pylori; oxidative stress

IN HUMAN TISSUES AND EXPERIMENTAL models of Helicobacter gastritis, reactive species play well-recognized roles in inflammation-induced toxicity to microbes and injury to the "bystander" gastric mucosa (12, 29, 50, 55, 56). In a variety of tissues, exposures to oxidants activate diverse intracellular signaling pathways (51). Among these signals are the intracellular accumulation of Ca²⁺ and Zn²⁺ in their labile forms (10, 17, 19, 49, 54). When uncontrolled, such divalent cation accumulation can exacerbate tissue injury (14, 42). When released in moderation, such signals may be protective (3, 48, 51). To more clearly explore the implications of such signals, it is important to clearly delineate their magnitude and temporal characteristics.

In this study, we used calcium ion reporters (fura-2 and

In this study, we explored the ability of one class of oxidants, the chloramines, to cause disturbances in intracellular homeostasis of Ca²⁺ and Zn²⁺ in epithelial cells of a primary functional unit, the secretory gland of gastric mucosa. The prototype in this class of oxidants, monochloramine (NH₂Cl), is produced through the reaction of neutrophil-derived hypochlorous acid (HOCl) with bacteria-derived ammonia (NH₃) (21, 53). Monochloramine is cell permeant and relatively stable in aqueous environments (20, 21). Molecular species that consume or neutralize the oxidant Cl· of chloramine species include (but are not limited to) glutathione and other peptides and proteins in which thiol (-SH) groups or clusters are integral to structure or enzymatic functions (11, 44). Recent studies have implicated such thiol groups in structural proteins and enzymes that regulate cell uptake and intracellular movements of Ca^{2+} and Zn^{2+} (1, 37, 38, 47). These considerations led us to hypothesize that exposure of gastric epithelial cells to NH₂Cl would elicit a distinct profile of disturbances in intracellular Ca²⁺ and Zn²⁺ through oxidation of thiol groups on proteins that bind and transport divalent cations.

Recently, we documented (10) such oxidant-induced disturbances in intracellular Ca2+ concentrations ([Ca2+]i) in epithelial cells of the colon mucosa, a specialized tissue in which NH₂Cl would be generated by the interaction of activated neutrophils and NH₃-producing bacteria. In the present study we used fluorescence-based reporters fura-2 and mag-fura-2 to examine, in detail, the mobilization of Ca²⁺ in parietal cells of the gastric gland during exposure to oxidants such as NH₂Cl. The parietal cell was chosen for study, in part, because its secretory dysfunction is an early consequence of acute, fulminant Helicobacter pylori gastritis (25, 40). In addition, the gastric parietal cell is highly enriched in mitochondria (27) and thus has the potential for use in evaluating cell energetics and classic pathways of apoptosis during oxidant-induced disturbances in $[Ca^{2+}]_i$ and intracellular Zn^{2+} concentration ([Zn²⁺]_i). Moreover, it is feasible to monitor the emptying and filling of intracellular Ca²⁺ stores in the gastric parietal cell with fluorescent reporters such as mag-fura-2 (31, 32). This approach offered the opportunity for a detailed examination of oxidant-induced disturbances in store emptying and repletion.

mag-fura-2) to monitor [Ca²⁺] in the cytoplasm and content within intracellular stores, respectively. It is well recognized that high-affinity calcium reporters are also responsive to heavy metals such as Zn^{2+} , often with much higher affinity (2). Thus, initially, we performed studies to demonstrate that a heavy

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metal chelator [N,N,N',N']-tetrakis-(2-pyridylmethyl)ethylenediamine, TPEN] may be used to screen out contributions from metal cations such as $\mathrm{Zn^{2+}}$, without impairing responses of the fluorescent reporters to calcium signals in the physiological range (100 nM to 1 μ M). After these initial validation studies, the primary goal of our studies was to determine the magnitude and time course of $\mathrm{Ca^{2+}}$ accumulation in response to NH₂Cl. A second goal was to determine the sources and possible mechanisms behind that accumulation. Our third goal was to determine whether the effectiveness of different antioxidant strategies would depend on the timing of their administration relative to the exposure to NH₂Cl.

MATERIALS AND METHODS

Gland isolation. Animal use protocols, including anesthesia, euthanasia, and experimental procedures, were approved by an independent review board in the Harvard Medical School, which oversees animal use at Brigham and Women's Hospital. New Zealand White rabbits (~2 kg) were anesthetized and underwent midline laparotomy. The aorta was cannulated and perfused retrograde with warmed (37°C) phosphate-buffered saline solution (mM: 150 NaCl, 0.6 NaH₂PO₄, 3 K₂HPO₄, 1 CaCl₂, and 1 MgCl₂ with 100 μM cimetidine, adjusted to pH 7.4 with HCl/NaOH). The gastric mucosa was separated from the underlying muscularis. Isolated glands were prepared with a modification (8) of previously published methods (4, 31). Collagenase type I (Sigma, St. Louis, MO) was used for ~60-min digestion with BSA in Dulbecco's modified Eagle's medium (DMEM, Sigma; with 100 μM cimetidine, pH 7.4). Glands were used within 8 h of isolation.

Dye loading. Fura-2 AM and mag-fura-2 AM (Molecular Probes, Eugene OR) were diluted in dimethyl sulfoxide to 1.0 mM. Glands were loaded at room temperature in DMEM (100 μM cimetidine, pH 7.4) with dye concentrations between 4 and 8 μM for 25 min. Subsequently, glands were rinsed several times in dye-free DMEM, mounted on glass coverslips, and then transferred to the microscope stage (Olympus IMT-2 or Nikon TE-2000), where they were perfused with Ringer solution (mM: 145 NaCl, 2.5 KH₂PO₄, 1.0 MgSO₄ or 1.0 MgCl₂, 1.0 CaCl₂, 10 HEPES, and 10 glucose, pH 7.4) at room temperature.

Gland permeabilization. In some cases, glands were permeabilized with digitonin (31, 32) after they had been mounted on coverslips and were under microscopic observation. After an initial rinse and stabilization in Ringer solution, glands were perfused with intracellular buffer [ICB; mM: 125 KCl, 25 NaCl, 0.3 CaCl₂, 0.5 MgCl₂, 10 HEPES, 0.5 Na₂ATP, and 0.5 ethylene glycol-bis(β-aminoethyl ether)-N,N,N',N' -tetraacetic acid (EGTA), pH 7.25, room temperature] containing 10 μM digitonin.

Imaging and ratiometric measurements. Simultaneous fluorescence measurements were obtained within each gland, in 6-10 individual cells. Fura-2- and mag-fura-2-loaded cells were excited alternately at 340 and 380 nm with a T.I.L.L. Photonics Polychrome IV system (Martinsried, Germany). Emitted light was collected at 520 \pm 15 nm after alternating excitation; the ratio of emission intensities provides an index of the Ca2+ concentration in the cytoplasm (fura-2) or subcellular compartments in permeabilized glands (mag-fura-2) (31, 32, 41). Digital images of glands were captured with a digital charge-coupled device camera (Hamamatsu ORCA-ER). Images were processed with compatible software (Universal Imaging, Downington, PA) to yield background-corrected pseudocolor images reflecting the 340 nm-to-380 nm ratio. Images were acquired every 10 s to minimize photobleaching. Contributions of autofluorescence were measured and taken into account, although these contributions were generally negligible because of bright staining of glands.

Preparation of monochloramine. Monochloramine (NH₂Cl) was prepared as described previously (10, 20, 21). Briefly, a 200- μ l

solution containing 500 mM NaOCl in water was added dropwise to 10 ml of 20 mM NH₄Cl and 5 mM Na₂HPO₄ in water at 0°C. This procedure resulted in a 5 mM NH₂Cl solution. Use of concentrated NH₂Cl solutions was completed within 6 h of preparation, as we observed that it remained stable in Ringer solution at concentrations ranging from 50 to 200 μ M with <10% loss of absorbance at 242 nm. The concentrated solution was kept on ice and then diluted to the final concentration just before each experiment. Taurine monochloramine (TaurNHCl) was generated under similar conditions by including taurine instead of NH₄Cl in the reaction mixture (20, 21). Concentrations were verified by measuring absorbance in a UV spectrophotometer at 242, 292, and 252 nm for NH₂Cl, HOCl, and TaurNHCl, respectively. [NH₂Cl], [HOCl], and [TaurNHCl] were then quantified with molar extinction coefficients reported previously (53).

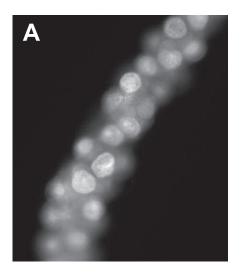
Western blot. Protein determinations were made with the Bradford protein assay (Sigma). SDS-PAGE was performed according to Laemmli (36). Samples were boiled for 5 min. For Western blotting, primary (mouse anti-human) antibodies were diluted in Tris-buffered saline-0.1% Tween 20 as follows: anti- sarco(endo)plasmic reticulum Ca²⁺-ATPase (SERCA)2 1:1,000 (clone C2C12, ABR-Affinity Bioreagents), anti-Na⁺/Ca²⁺ exchanger (NCX) 1:1,000 (clone 2A7-A1, ABR-Affinity Bioreagents), and anti-Na⁺-K⁺-ATPase, α_{1c}-subunit, 1:5,000 (Chemicon). Goat anti-mouse secondary antibody was used at 1:2,000 dilution. Blocking of nitrocellulose was done in 3% nonfat milk in Tris-buffered saline. Horseradish peroxidase was detected by chemiluminescence (LumiGLO, KPL; www.KPL.com), and the signal was visualized on Kodak Bio-Max X-ray film.

Additional reagents and methods. Nigericin, ionomycin, digitonin, DTT, and ascorbic acid (vitamin C, VitC) were obtained from Sigma. TPEN was obtained from Molecular Probes. For Ringer solution and ICB, calculations of free and bound concentrations of Ca²⁺, Zn²⁺, TPEN, and EGTA were performed with the internet-based WEBMAXSTANDARD program (http://www.stanford.edu/%7Ecpatton/webmaxcE.htm).

Data summary and statistical analysis. Fluorescence intensities were monitored at 10-s intervals throughout each experiment. At discrete time intervals, measurements were summarized as means \pm SE. For comparison between treatments, unless stated otherwise, measurements in different regions of interest (6–10 cells for each gland) were combined to provide a single integrated value at each time point for each gland. Unless stated otherwise, comparisons were performed by analysis of variance (ANOVA) for sequential measurements performed with purchased software (Sigma Stat, version 2.0, Jandel).

RESULTS

Measurements of $[Ca^{2+}]_i$ *in gastric with fura-2: calibration* and control for contributions of heavy metal polyvalent cations. Previous reports have demonstrated the feasibility of using dyes such as fura-2 or fluo-3 to monitor [Ca²⁺]_i in gastric parietal cells in response to physiological stimuli (9, 41). These approaches depend on assumptions (41) that have not been fully tested under conditions in which glands might be exposed to oxidants or other potentially cytotoxic agents. One assumption is that the dye is concentrated in the cytoplasm. To confirm loading of dye in the cytoplasm, glands loaded with fura-2 AM were monitored during excitation at 340 and 380 nm. As has been noted previously (41), these cells are easily distinguished from chief cells and other cell types by their characteristic large size and bulging basolateral surface, often associated with a conical configuration (Fig. 1), before and after exposure of glands to 10 µM digitonin, an agent that permeabilizes the cell membrane without disturbing transport of secretory function of intracellular organelles (8, 28, 31). In five separate experi-



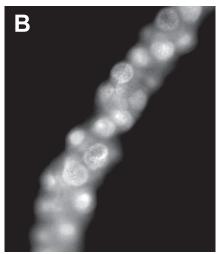


Fig. 1. Digital images in gray scale of an isolated rabbit gastric gland loaded with fura-2 AM (8 μM) for 25 min. A: fluorescence excitation at 340 nm when perfused with Ringer solution. B: fluorescence excitation when perfused with Ringer solution containing free N,N,N,N'-tetrakis-(2-pyridylmethyl)ethylenediamine (TPEN) $\sim\!20~\mu M$. Original magnification $\times 30$.

ments, exposure to digitonin markedly decreased fluorescence at both wavelengths by 85–90% (data not shown), indicating that the dominant contribution to fluorescence signals comes from the cytoplasm.

A second assumption is that contributions of interfering cations are not detected by the fluorescent reporter. We performed two sets of studies to explore the potential influence of heavy metal cations on fura-2 fluorescence in gastric parietal cells. First, we determined the apparent K_d of fura-2 for Zn^{2+} within the parietal cell of the gastric gland. Gastric glands loaded with fura-2 AM were perfused with Ringer solutions containing a strong chelator (0.3 mM EGTA) and no added Ca²⁺. The glands were then exposed to solutions containing the Zn²⁺ ionophore pyrithione (50 µM) and progressively higher [Zn²⁺] (0.25, 2.5, 5.0, 7.5, 10, and 25 nM). As shown in a typical recording (Fig. 2), the dye is responsive over the range of 0.25-10 nM. In five glands (n = 32 cells) we determined that the apparent K_d of fura-2 for Zn^{2+} is 2.9 ± 0.3 nM in the parietal cell. These findings indicate that physiologically and pathologically relevant increases of [Zn²⁺]; in the nanomolar range may well be reported by fura-2.

In the second set of studies, we explored the feasibility of using the heavy metal chelator TPEN to buffer against increases in heavy metals without compromising the quality of the calcium signals. This heavy metal chelator has very high affinity for non-Ca²⁺ heavy metals such as Zn²⁺ and Fe²⁺ ($\sim 10^{-15}$ M) (2) and low affinity for Ca²⁺ ($K_d \sim 100~\mu$ M) (2, 8). Thus exposure to TPEN in the range of $10-20~\mu$ M would screen out contributions of interfering metal cations while permitting fura-2 to respond to physiological levels of Ca²⁺ accumulation in the cytoplasm (100 nM to 10 μ M) (30). As shown in Fig. 1 the presence of 20 μ M TPEN alters baseline patterns of fura-2 fluorescence minimally or not at all.

Initially, we determined the apparent K_d of fura-2 for Ca²⁺ when loaded in the parietal cell. As above, gastric glands loaded with fura-2 AM were perfused with Ringer solutions containing a strong chelator (0.3 mM EGTA) and no added Ca²⁺. The glands were then exposed to solutions containing the Ca²⁺ ionophore ionomycin and TPEN (20 μ M); after equilibration with ionomycin glands were exposed to progressively higher [Ca²⁺] (100 nM to 20 μ M). As shown in Fig. 3 exposure of glands to incremental increases in [Ca²⁺] in the

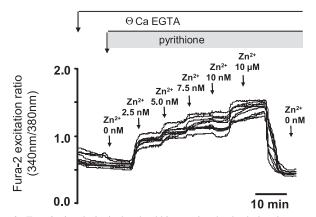


Fig. 2. Fura-2 signals in isolated rabbit gastric glands during incremental increases in Zn²⁺ concentration ([Zn²⁺]). Glands were exposed initially to Ca²⁺-depleted Ringer solution (no added Ca²⁺, 0.5 mM EGTA) and then to EGTA/Ringer with [Zn²⁺] varying from 2.5 nM to 10 μ M, in the presence of pyrithione (50 μ M). From averaged values for 4 glands, the K_d of fura-2 for Ca²⁺ is 2.9 \pm 0.3 nM.

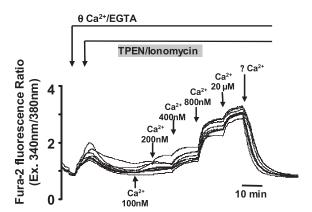


Fig. 3. Fura-2 signals in isolated rabbit gastric glands during incremental increases in Ca²⁺ concentration ([Ca²⁺]). Glands were exposed initially to Ca²⁺-depleted Ringer solution (no added Ca²⁺, 0.5 mM EGTA) and then to EGTA/Ringer with [Ca²⁺] varying from 100 to 800 nM, in the presence of ionomycin (10 μ M) and TPEN (20 μ M). From averaged values for 8 glands, the K_d of fura-2 for Ca²⁺ is 380 \pm 35 nM.

presence of ionomycin and TPEN disclosed incremental and reversible increases in the excitation ratio, thereby permitting direct correlation of fura-2 signals to $[Ca^{2+}]$. Under these conditions, we calculate that the apparent K_d of fura-2 for Ca^{2+} is 380 ± 35 nM (mean \pm SE; n=49 cells in 9 glands). This value is higher than that reported in cultured cells (22, 26) but close to that reported previously for parietal cells in gastric glands (41). Based on a log-linear plot of fluorescence vs. concentration, baseline levels of $[Ca^{2+}]_i$ in Ringer-perfused glands are ~ 160 nM.

We also evaluated whether the presence of TPEN would interfere with the ability of fura-2 to monitor Ca2+ accumulation in the cytoplasm in the parietal cell during maneuvers known to release Ca²⁺ from intracellular stores. In one set of studies, glands were exposed to ionomycin (10 µM) and nigericin (7 µM) together, maneuvers that not only optimize release of Ca²⁺ but also may release loosely bound heavy metal cations from acidic as well as nonacidic intracellular stores (16, 23, 46). In seven individual gland experiments (Fig. 4), exposure to ionomycin and nigericin increased the fluorescence excitation ratio (340/380 nm), indicating release of Ca²⁺. The experiments were also performed when glands were preperfused with Ca²⁺-depleted Ringer containing 20 µM TPEN, a level sufficient to bind heavy metals accumulating in the low micromolar range (2). Comparing outcomes (peak effects) in studies performed in the presence or absence of TPEN (Fig. 4, bottom), it appears that the presence of TPEN

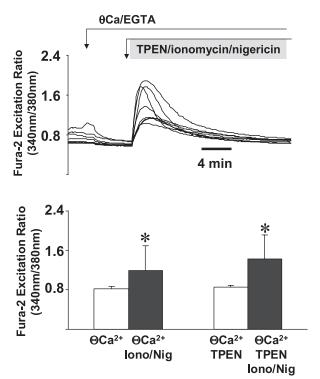
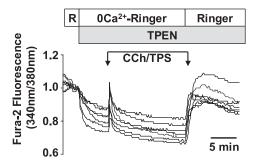


Fig. 4. Summary of measurements of fura-2 fluorescence during release of Ca^{2+} and other metals during exposure to ionomycin (10 μ M) and nigericin (7 μ M) alone or in the presence of the heavy metal chelator TPEN. *Top*: individual recording showing responses of 7 parietal cells in a single gastric gland. *Bottom*: summary of peak responses in glands exposed to ionomycin/nigericin in the presence or absence of TPEN (n=7/group). Results are means \pm SE. *P<0.0001, comparing measurements before and at the peak response to ionomycin/nigericin (paired t-test).



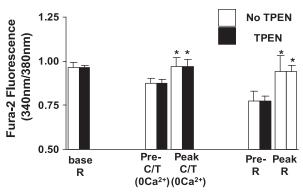


Fig. 5. Effect of TPEN on "pure" intracellular Ca^{2+} signals monitored by fura-2. *Top*: individual experiment, each line representing a recording from an individual parietal cell. The recordings begin when the gland is exposed to Ringer solutions containing no added Ca^{2+} , 0.5 mM EGTA and TPEN (20 μ M). The presence of TPEN had no effect on signals generated during exposure to 100 μ M carbachol (CCh) and 2 μ M thapsigargin (TPS). Also, TPEN had no effect on signals when extracellular Ca^{2+} (1 mM) was restored. *Bottom*: summary of changes in fura-2 signals under control conditions (n=7) or in the presence of TPEN (n=6). *Differences between "pre" and "peak" signals are significant (P<0.01) within each group, but not between the 2 groups of glands. base R, baseline Ringer solution; C/T, carbachol/thapsigargin; pre-R, prerestoration of Ringer solution; peak R, peak after restoration of Ringer solution.

did not substantially alter the fluorescence excitation ratio (340/380 nm) at baseline or during exposure to ionomycin/nigericin.

We then determined whether TPEN would interfere with the ability of fura-2 to measure physiological increases in [Ca²⁺]. Glands were perfused with Ringer solutions under control conditions and then during exposure to a combination of carbachol and thapsigargin. These agents cause release of Ca²⁺ from intracellular stores and prevent reuptake, thereby maximizing accumulation in the cytoplasm (30-32). In addition, irreversible release of intracellular stores activates capacitative entry of Ca²⁺ from the extracellular spaces to the cytoplasm, increasing the magnitude of Ca²⁺ accumulation (43). Studies were thus performed under control conditions and during exposure to solutions in which total Ca²⁺ and total TPEN were calculated to maintain [Ca²⁺] at 1 mM and free TPEN at \sim 20 μ M (assuming $K_{\rm d}$ of TPEN for Ca²⁺ of \sim 60–100 μ M, total Ca²⁺ 1.2 mM, total TPEN 220 μ M). As shown in Fig. 5, the presence of TPEN did not alter [Ca²⁺] signals elicited by carbachol/thapsigargin or during reentry from the extracellular fluid. These studies confirm that TPEN screens out contributions of heavy metals but does not interfere with the ability of fura-2 to monitor accumulation of Ca²⁺ in the cytoplasm.

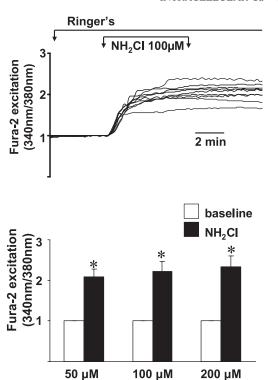


Fig. 6. Measurements of fura-2 signals during exposure of gastric glands to NH₂Cl. *Top*: recordings from individual parietal cells (n=7) in an individual gland exposed to 100 μ M NH₂Cl. Recording begins with the gland perfused by standard Ringer solution, then exposure to NH₂Cl, and then standard Ringer solution. Note absence of reversibility. *Bottom*: summary of responses (steady-state response after 12 min) to NH₂Cl at different doses (50, 100, and 200 μ M; n=5 experiments at each concentration). All experiments were performed with the protocol outlined at *top*. Results are means \pm SE of responses of individual glands.*P<0. 001 compared with baseline (ANOVA).

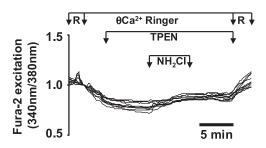
Responses of fura-2-loaded gastric glands to NH_2Cl . Previously we noted (10) that NH_2Cl remained stable for 6 h in Ringer solution with <10% loss of absorbance at 242 nm. All experiments were conducted within this interval after preparation of NH_2Cl . In initial experiments, fura-2 responses were monitored during exposure of isolated gastric glands to Ringer solutions with $[NH_2Cl]$ of 50, 100, or 200 μ M. Initially, exposure to NH_2Cl led to a marked increase, then a more prolonged and gradual increase in the fluorescence excitation ratio 340/380 nm (Fig. 6, *top*). These signals were not reversed when the oxidant was removed. Curiously (Fig. 6, *bottom*), minimal differences were observed in responses to increasing $[NH_2Cl]$ over the range tested.

Control studies were performed to evaluate the specificity of the responses to NH₂Cl. Similar effects were not observed when glands (n=3 in each group) were exposed to NH₄Cl (20 mM), HOCl (200 μ M), H₂O₂ (200 μ M) or the membrane-impermeant compound TaurNHCl. These findings indicate that responses are confined to membrane-permeant choramine species.

Contributions of intracellular stores of Ca^{2+} in response to NH_2Cl . In other cell types, heavy metal cations such as Zn^{2+} have been shown to contribute significantly to the fura-2 signal observed during exposure to NH_2Cl (10). We performed studies to monitor the effects of NH_2Cl on release of Ca^{2+} from intracellular sources in the absence of interfering metal cations.

Isolated fura-2-loaded gastric glands were perfused with Ca^{2+} free Ringer solution containing EGTA (0.5 mM) and TPEN (20 μ M). Responses were then monitored before, during, and after NH₂Cl was present at a concentration of 50, 100, or 200 μ M. When glands are perfused under these conditions, the excitation ratio decreases markedly (Fig. 7), corresponding to levels of [Ca²⁺] of 10 nM or lower. Starting at this baseline, NH₂Cl elicits dose-dependent, modest, and nonreversible increases in [Ca²⁺]_i. With the calibrations in Fig. 3 as a reference, exposure to 200 μ M NH₂Cl increases [Ca²⁺]_i to levels between 200 and 300 nM above the baseline.

To more conclusively evaluate effects of NH₂Cl on emptying and filling of intracellular stores, we performed studies in isolated rabbit gastric glands loaded with mag-fura-2 AM. This low-affinity, Ca^{2+} -sensitive reporter ($K_d \sim 100 \mu M$) has been used to monitor [Ca2+] in subcellular compartments of the gastric gland after permeabilization to eliminate dye in the cytoplasm (31, 32). At baseline, the dominant portion of the signal was attributable to the high content of Ca²⁺ in the endoplasmic reticulum (31, 32). The very low affinity of mag-fura-2 for $\mathrm{Mg^{2+}}$ ($K_{\mathrm{d}} \sim 1.5$ mM) makes it very unlikely that $\mathrm{Mg^{2+}}$ contributes to mag-fura-2 fluorescence in a store that has a high content of Ca²⁺, for which mag-fura-2 has a much higher affinity ($K_d \sim 100 \mu M$) (26). Moreover, studies in this same preparation of the isolated rabbit gastric gland (31, 32), and in other cell types and species (18), also excluded a significant contribution of Mg²⁺ to mag-fura-2 fluorescence in permeabilized preparations.



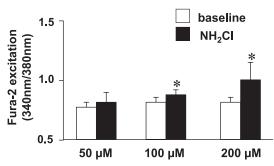


Fig. 7. Top: Recordings from individual parietal cells (n=8) in an individual gland exposed to $100~\mu M$ NH₂Cl in the presence of TPEN ($20~\mu M$). Recording begins with the gland perfused by standard Ringer solution (R), then switch to calcium-depleted Ringer solution, and then exposure to NH₂Cl and then standard Ringer solution. Note absence of reversibility. Bottom: averaged responses to NH₂Cl at different doses (50, 100, and $200~\mu M$). Each bar represents responses in 4 or 5 separate experiments (n=6-9 parietal cells/gland), each conducted in exactly the same sequence as described at top. Results are means \pm SE, with y-axis indicating excitation ratio ($340/380~\rm nm$). * $P<0.01~\rm compared$ with baseline by ANOVA.

However, mag-fura-2 also has a much higher affinity for heavy metals such as Zn^{2+} ($K_d \sim 3$ nM) (26). To eliminate contributions of heavy metals, studies were again performed in the presence of TPEN at a concentration (10 µM) high enough to chelate heavy metal components but not so high as to alter Ca²⁺ content in highly concentrated intracellular stores (6, 30). In six permeabilized glands loaded with mag-fura-2 AM, exposure to 10 µM TPEN alone modestly increased (<10% above baseline) the fluorescence excitation ratio (340/380 nm), indicating that intracellular stores monitored by mag-fura-2 may contain small amounts of interfering non-Ca²⁺ metal cations that quench the signal. Further increases were not observed when TPEN concentration was increased to 20 µM (n = 3; data not shown). An additional maneuver to confirm that mag-fura-2 reports Ca²⁺ in the high-content store was to expose permeabilized glands to thapsigargin, which inhibits SERCA, thereby preventing Ca²⁺ reuptake into the highcontent store. As reported by Hofer and Machen (31, 32), the store monitored by mag-fura-2 was depleted by application of 2 μ M thapsigargin (n = 3).

To evaluate the effects of NH₂Cl on intracellular stores of Ca²⁺, we exposed digitonin-permeabilized gastric glands loaded with mag-fura-2 to solutions containing 20 μ M TPEN and 100 μ M NH₂Cl. As shown in Fig. 8, rapid decreases in the excitation ratio were observed, demonstrating that exposure to NH₂Cl causes a marked depletion of intracellular pools of Ca²⁺ within the parietal cell. Similar responses were observed in six other glands, confirming a nonreversible depletion of intracellular Ca²⁺ stores in response to NH₂Cl, 68 \pm 4% compared with baseline (P < 0.001).

Contributions of extracellular Ca^{2+} . We next evaluated the effects of NH₂Cl on influx of extracellular Ca^{2+} to the cytoplasm. As noted above, removal of Ca^{2+} from perfusates leads to significant depletion of Ca^{2+} in the cytoplasm, indicating influx from extracellular sources under resting conditions. Two sets of studies were performed to evaluate whether exposure to NH₂Cl accelerates Ca^{2+} influx. In the first set, we compared

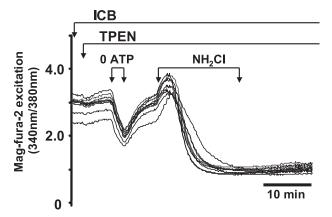


Fig. 8. Mag-fura-2 signals during exposure of an isolated, permeabilized gastric gland to NH₂Cl in the presence of TPEN. Mag-fura-2-loaded glands were permeabilized with α -toxin and then mounted on a coverslip for fluorescence imaging at 520 nm (alternating excitation at 340 and 380 nm). Each line represents a recording of mag-fura-2 responses in an individual parietal cell. Recording begins as gland is perfused with intracellular buffer (ICB; 0.5 mM ATP) and then exposed to TPEN (20 μM). A brief removal and repletion of ATP from the perfusate confirms that the gland is permeabilized. Exposure to 100 μM NH₂Cl demonstrates complete and nonreversible emptying of intracellular stores.

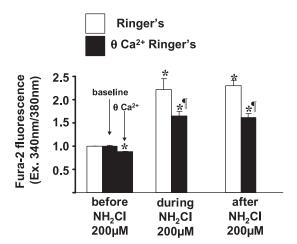


Fig. 9. Comparison of responses to NH₂Cl in the presence or absence of extracellular Ca²⁺. Each column represents integrated response (mean \pm SE) of 6 glands (6–8 cells/gland) when a stable baseline was achieved after the change in experimental condition. *P < 0.05 compared with Ringer solution baseline (ANOVA); ¶P < 0.05 compared with baseline or Ringer solution control

responses to NH₂Cl in Ringer solution or to NH₂Cl in Ca²⁺-free Ringer solutions. In these studies, TPEN was not included, so as not to create undue disturbances by the exclusion of heavy metals. As summarized in Fig. 9, increases in excitation ratio were significantly lower (P < 0.01) when glands were exposed to 200 μ M NH₂Cl in the absence of extracellular Ca²⁺. These experiments confirm that extracellular Ca²⁺ plays a significant role in the overall response to NH₂Cl.

We then performed studies to more precisely evaluate whether influx of Ca2+ is accelerated during the exposure to NH₂Cl. The strategy in these studies was to observe Ca²⁺ influx after depletion and then restoration of extracellular Ca²⁺, after exposure to NH₂Cl. After equilibration with Ringer solutions, two groups of glands were perfused for 5–7 min with Ca²⁺-free Ringer solution containing 20 μM TPEN. The control group was perfused for an additional 5-7 min with Ca²⁺-free/TPEN solutions, while the other group was perfused with Ca²⁺-free/TPEN containing 200 µM NH₂Cl. Both groups were then perfused with standard 1 mM Ca²⁺-Ringer solutions containing total Ca²⁺ (1.2 mM) and total TPEN (220 µM) sufficient to maintain free levels of TPEN at ~20 μM. As shown in Fig. 10, removal of extracellular Ca²⁺ in the presence of TPEN elicited significant and comparable decreases in [Ca²⁺]_i in both groups of glands. As noted above (Fig. 7), exposure to NH₂Cl elicited a modest increase in [Ca²⁺]_i due to release of intracellular stores. When extracellular Ca²⁺ was restored, increases in [Ca²⁺]_i were observed in both groups. The magnitude of the recovery in the NH₂Cl group was significantly higher than in the control group, arguing that exposure to NH₂Cl accelerates Ca²⁺ influx above baseline levels.

To provide additional evidence that this accelerated entry from the outside might be due to store-operated entry, influx of Ca²⁺ during the restoration of extracellular Ca²⁺ was monitored after exposure to NH₂Cl. Studies were performed under control conditions or in the presence of 2-aminoethoxydiphenyl borate (2-APB), which has been shown to block store-operated influx of Ca²⁺ in other cell types (5). As shown in

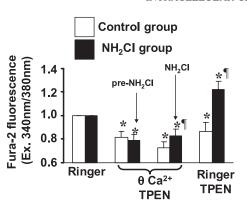


Fig. 10. Influx of extracellular Ca^{2+} after exposure to NH_2Cl . Both groups of glands (control and NH_2Cl) were initially perfused with Ca^{2+} -free Ringer/ TPEN for 5 min, then with Ca^{2+} -free/TPEN alone or Ca^{2+} -free/ TPEN + NH_2Cl for 5 min, and then with Ringer/TPEN for 20 min. Because Ringer solution contains 1 mM Ca^{2+} , TPEN concentration was adjusted so that free [TPEN] was $\sim 20~\mu$ M. Each column represents integrated response (mean \pm SE) of 6 glands (6–8 cells/gland). *P<0.05 compared with Ringer solution baseline (ANOVA); $\PP<0.05$ compared with control.

Fig. 11, 2-APB was added to the perfusate after exposure to 100 μ M NH₂Cl and just before restoration. In a set of four paired experiments, glands from the same harvest were studied sequentially to minimize variation in responses to NH₂Cl. During the Ca²⁺ restoration phase (after exposure to NH₂Cl), the increase in fluorescence excitation ratio was significantly reduced in the presence of 2-APB (Δ 15 min after restoration: 0.17 \pm 0.03 arbitrary units) compared with that in its absence (0.34 \pm 0.06; P < 0.05, ANOVA multiple comparisons). In control studies conducted in the absence of NH₂Cl, responses to removal and restoration of extracellular Ca²⁺ were not altered in the presence of 2-APB (data not shown). Thus it appears that a component of NH₂Cl-induced influx of Ca²⁺ is attributable to store-operated entry.

 NH_2Cl -induced Ca^{2^+} signals in presence of antioxidants. To determine whether pretreatment with antioxidants would prevent responses, glands loaded with fura-2 were perfused with Ringer solutions containing antioxidants (1 mM VitC, 100 μ M DTT, or 1 mM DTT) before exposure to NH_2Cl . With each antioxidant (n=4 or 5 glands), pretreatment for 4–8 min prevented any response to NH_2Cl at concentrations up to 200 μ M (data not shown).

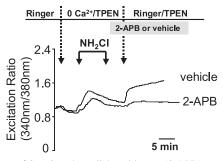


Fig. 11. Effects of 2-aminoethoxydiphenyl borate (2-APB) on store-operated entry induced by NH₂Cl. Superimposed recordings from fura-2-loaded glands from the same harvest are shown. Both glands were exposed to $100~\mu M$ NH₂Cl in a protocol similar to that in Fig. 9, with similar responses. As before, $20~\mu M$ TPEN was present to screen out contributions from heavy metal cations. After removal of NH₂Cl, glands were exposed to 2-APB or vehicle (ethanol 1:1,000), and then extracellular Ca^{2+} was restored.

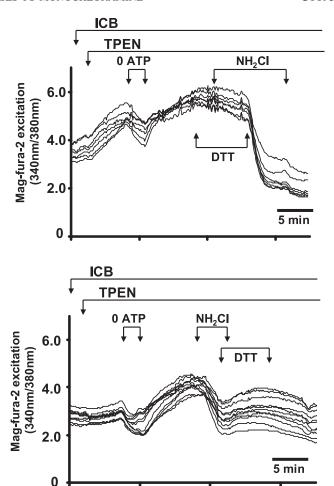


Fig. 12. Thiol reduction as a means of preventing or reversing NH₂Cl effects on intracellular Ca²⁺ stores. Glands loaded with mag-fura-2 were permeabilized with *Staphylococcus aureus* α -toxin in ICB and then mounted on the microscope stage. Before exposure to NH₂Cl, each gland was perfused with ICB from which ATP was removed, to confirm successful permeabilization. *Top*: recording in an individual gland exposed to NH₂Cl in ICB already containing 1 mM dithiothreitol (DTT). DTT is then removed, with NH₂Cl remaining present, thereby allowing stores to empty. *Bottom*: recording in an individual gland exposed to 1 mM DTT after initiation of responses to 100 μ M NH₂Cl. NH₂Cl is then removed, with DTT remaining, thereby arresting and partially reversing effects of NH₂Cl. In both panels, individual lines represent recordings from individual parietal cells.

Of greater interest was whether responses might be arrested and reversed if the antioxidant were administered during ongoing exposure to NH₂Cl. To characterize the influence of different antioxidants on Ca²⁺ release and filling of intracellular stores, glands were loaded with mag-fura-2 AM, permeabilized with α -toxin, and equilibrated with TPEN-containing intracellular buffer. As shown in Fig. 12, *top*, exposure to DTT before application of 200 μ M NH₂Cl prevents depletion of the stores. Moreover, after NH₂Cl has been added, application of DTT arrests and partially reverses the precipitous decrease in store content (Fig. 12, *bottom*). In contrast, pretreatment with VitC prevented emptying of stores during subsequent exposure to NH₂Cl. However, application of VitC after exposure to NH₂Cl did not lead to any reversal of store depletion (n=4 individual glands in separate experiments; data not shown).

We then performed studies to provide insight into the effects of different antioxidants during NH₂Cl-induced accumulation

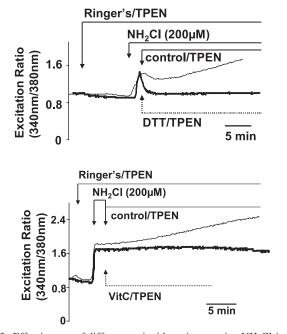


Fig. 13. Effectiveness of different antioxidants in reversing NH₂Cl-induced accumulation of Ca^{2+} in the cytoplasm. Before exposure to NH₂Cl, each gland was perfused with Ringer solution containing a sufficient amount of TPEN (0.2 mM) and Ca^{2+} (1.2 mM) to maintain free [TPEN] at ~20–50 μ M. *Top*: recording in individual glands (from same harvest) exposed to NH₂Cl in the presence or absence of the thiol-reducing agent DTT. Thin line represents summary of 8 parietal cells in an individual gland exposed to NH₂Cl alone; thick line represents summary of 7 parietal cells in an individual glands (from same harvest) exposed to NH₂Cl and DTT. *Bottom*: recording in individual glands (from same harvest) exposed to NH₂Cl in the presence or absence of the oxidant scavenger vitamin C (VitC). Thin line represents summary of 8 parietal cells in an individual gland exposed to NH₂Cl alone; thick line represents summary of 8 parietal cells in an individual gland exposed to NH₂Cl alone; thick line represents summary of 8 parietal cells in an individual gland exposed to NH₂Cl and VitC.

of Ca²⁺ in the cytoplasm of the parietal cell. Glands were loaded with fura-2 AM, equilibrated with Ringer solution containing total TPEN (220 µM) and Ca²⁺ (1.2 mM) to maintain free [TPEN] at 20-50 μM, and then exposed to NH₂Cl (200 µM). In earlier studies (10), we observed that addition of DTT (up to 1 mM) does not destabilize or deplete NH₂Cl levels in the Ringer solution. Thus it was possible to observe the effects of DTT in the presence of an ongoing exposure to NH₂Cl. As shown in Fig. 13, top, addition of 1 mM DTT to the perfusate at the peak of the NH₂Cl effect led to a rapid decrease in [Ca²⁺]_i, consistent with the previous observation that addition of DTT would reverse NH₂Cl-induced depletion of intracellular Ca²⁺ stores (Fig. 12) and suggesting, in addition, that store-operated entry of extracellular Ca²⁺ is also reversed. The rate of reversal was measured as the decline from peak value at 1 min. This rate was significantly accelerated after addition of DTT (P < 0.01): 43.2 \pm 6.5% (n = 8glands) compared with $9.1 \pm 1.9\%$ (n = 10 glands) after simple exchange of the perfusing Ringer solution.

In contrast, we observed that addition of 1 mM VitC consumed NH₂Cl in Ringer solutions, as monitored by a complete loss of absorbance at 242 nm. Thus it was not feasible to observe the effects of VitC while maintaining a steady level of NH₂Cl in the perfusate. In studies of this antioxidant, glands were exposed briefly to Ringer solution containing NH₂Cl (200 μ M) and then to Ringer solution or Ringer solution containing

VitC (1 mM). As shown in Fig. 13, *bottom*, introduction of VitC did not reverse the effects of NH₂Cl. In the presence of VitC (n = 7), we never observed a continued accumulation of Ca²⁺ in the cytoplasm that was occasionally present in control glands (3 of 7) after removal of NH₂Cl.

Finally, we performed Western blots to evaluate effects of NH₂Cl on expression of Ca²⁺ transport proteins. We used an antibody directed at SERCA2, which is responsible for store refilling (7) and is expressed in gastric mucosa (39). In addition, we utilized an antibody directed at the plasma membrane NCX, which has been implicated in disposal of cytoplasmic Ca²⁺ and isoforms of which have been identified in intestinal epithelia (15). As shown in Fig. 14, top, incubation of gastric glands for 1 h with NH₂Cl (100 µM) almost completely depleted expression of the SERCA2 isoform at its characteristic ~110-kDa position, an effect prevented by pretreatment with 1 mM DTT. Other, nonspecific bands were also detected and were not responsive to NH₂Cl or rescue with DTT. Similarly, the characteristic bands (15) for NCX (120 and 70 kDa) were depleted in response to NH₂Cl and protected when DTT was present. Additional control studies (not shown) demonstrated that pretreatment with DTT, in the absence of NH₂Cl, had no effect on expression of SERCA2 or NCX. Moreover, we found that NH₂Cl had no effect on expression on expression of the α -subunit of the Na⁺-K⁺-ATPase, indicating that the effects of NH₂Cl are not attributable to nonspecific destruction of transporter proteins.

DISCUSSION

Considerable experimental evidence indicates that NH₂Cl is likely to be present in gastric mucosa during acute *H. pylori*-induced gastritis. When luminal solutions contain relevant concentrations of NH₃, mucosal injury is amplified under conditions that activate neutrophils (12, 50). In addition, strategies aimed at neutralizing chloramines appear to attenuate mucosal injury caused by *Helicobacter* infestation (33). Accu-

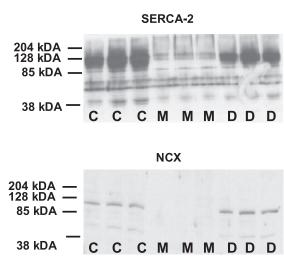


Fig. 14. Alterations in Ca^{2+} transporter expression during exposure to NH_2Cl . Western blots utilizing antibodies reacting with sarco(endo)plasmic reticulum Ca^{2+} -ATPase (SERCA; *top*) isoform 2 and with known isoforms for Na^{+}/Ca^{2+} exchangers (NCX; *bottom*) are shown. *Lanes C*, control glands incubated in Ringer solution for 1 h; *lanes M*, gland preparations incubated for 1 h in the presence of 100 μ M NH_2Cl ; *lanes D*, glands incubated for 1 h with both NH_2Cl (100 μ M) and DTT (1 mM).

mulation of NH_2Cl is not unique to the gastric mucosa: it can occur in colon and other organs where activated neutrophils come into contact with NH_3 generated from bacterial or host metabolism (20, 21, 24, 52). Thus the effects of chloramines on cell signaling pathways are of interest in cell systems potentially susceptible to different forms of oxidative and nitrosative stress.

Our studies indicate that, at pathologically relevant concentrations (20, 21, 24, 52), exposure to NH₂Cl causes accumulation of free Ca²⁺ in the cytoplasm of the gastric parietal cell. Accumulation of Ca²⁺ results from emptying of intracellular stores and from the ensuing activation of store-operated entry from the extracellular spaces. In addition, the effects of NH₂Cl on expression of SERCA2 and NCX argue that an additional effect of NH₂Cl is to impair mechanisms that dispose of Ca²⁺ loading in the cytoplasm. These effects are not duplicated by precursor oxidants, such as peroxide or HOCl, nor are they elicited by membrane-impermeant chloramines such as taurochloramine. Thus the effects of NH₂Cl on Ca²⁺ mobilization and accumulation in the cytoplasm appear to be specific to the chloramine moiety. These effects are neutralized by concurrent exposure to oxidant scavengers such as VitC; moreover, they can be prevented, arrested, and at least partly reversed by exposure to thiol-reducing agents such as DTT. This latter observation provides evidence that at least some substantial component of NH₂Cl-induced accumulation of Ca²⁺ is the result of a targeted and reversible oxidative process, and not an irreversible leakage of Ca2+ from injured compartments and cells. To our knowledge, these studies are the first to examine the effects of thiol oxidants on intracellular Ca²⁺ homeostasis in gastric epithelial cells in a primary cell preparation, the gastric gland. In addition, these are the first studies to directly evaluate the effects of pathologically relevant thiol-directed oxidants and reducing agents on release of Ca2+ from intracellular stores in any primary epithelial cell preparation, either functionally or at the molecular level.

These studies raise three issues for discussion. The first issue involves the mechanisms by which NH2Cl causes accumulation of Ca²⁺ within the cytoplasm of the gastric parietal cell. With respect to [Ca²⁺]_i, our studies indicate that NH₂Clinduced increases are due in part to release from intracellular stores. The comparison of responses to perfusates with and without Ca²⁺ makes it clear that entry of extracellular Ca²⁺ is also responsible for a major component of the NH₂Cl-induced signal. These responses are attributable to thiol modification of Ca²⁺ transport processes, since DTT prevented and reversed effects of NH₂Cl on fura-2 signals. Entry of extracellular Ca²⁺ might be due to influx through capacitative entry, activated when carbachol- and thapsigargin-sensitive stores are released (30–32). Consistent with this explanation is our observation that 2-APB at least partially inhibits influx of NH₂Cl-induced extracellular Ca²⁺. Further studies would be required to address the possibility that disposal of accumulating Ca²⁺ is also impaired by exposure to thiol-directed oxidants. Specific mechanisms to be studied would include the capacity of the mitochondrion to buffer the cytoplasm against increases in [Ca²⁺]_i (45) and clearance of Ca²⁺ in the cytoplasm through membrane extrusion processes, such as the plasma membrane Ca²⁺-ATPase (PMCA) (7).

The second issue raised by our studies involves the consequences of oxidant-induced increases in $[Ca^{2+}]_i$ and $[Zn^{2+}]_i$ in

the gastric gland. In diverse in vitro cell culture models, agonist-stimulated increases in [Ca²⁺]_i enhance mitochondrial respiration and substrate utilization (34, 45). In excess, increases in [Ca²⁺]_i induce mitochondrial depolarization and the initiating events of apoptosis and/or cell necrosis (42). However, the effects of more limited and controlled oxidantinduced increases in [Ca²⁺]_i, such as those reported here, have not been characterized. Indirect evidence suggests that such limited increases in [Ca²⁺]_i oppose the generally suppressive effects of the increases in [Zn²⁺] induced by NH₂Cl (Dubach JM, Naik HB, Beshire MA, Wieland AM, Walsh BM, Soybel DI, unpublished observations). Moreover, they also oppose the generally toxic effects of NH₂Cl-induced accumulation of Zn²⁺ in the cytoplasm. Thiol oxidant-induced alterations in the balance between [Ca²⁺]_i and [Zn²⁺]_i may thus prove to be a useful target in modulating injury caused by the inflammatory microenvironment caused by ischemia-reperfusion or infestation with *H. pylori*.

The last issue for discussion is technical, namely, the strengths and limitation of fluorescence methods used to monitor intracellular divalent cation signals. Calcium-sensing fluorescent indicator dyes all are responsive to other polyvalent cations. Potentially interfering polyvalent cations that might be released from intracellular pools include Fe²⁺, Fe³⁺, Cu²⁺, and Zn²⁺. The most widely used fluorescent reporter, fura-2, is quenched in response to Cu²⁺, Fe²⁺, and Fe³⁺. However, it responds to Zn²⁺ in a manner similar to Ca²⁺. Previous reports and manufacturer specifications indicate that, in vitro, fura-2 has a higher affinity for Zn²⁺ ($K_{\rm d} \sim 3$ –15 nM) than for Ca²⁺ ($K_{\rm d} \sim 150$ –300 nM) (26, 35, 41). Our calibration studies confirm these ranges of sensitivities to Zn²⁺ and Ca²⁺ in situ in cells of the rabbit gastric gland.

Mag-fura-2 has a much lower affinity for Ca^{2+} , with a reported K_d of 25–60 μ M (26), a property that has made it useful for monitoring intracellular stores of Ca^{2+} in the range of 10–100 μ M (31, 32). In cell-free solutions, mag-fura-2 has a reported K_d for Zn^{2+} of ~20 nM (13, 26). In situ, however, mag-fura-2 does not respond sensitively to $[Zn^{2+}]$ in the nanomolar range, perhaps because responses to high-concentration, intracellular Ca^{2+} stores are dominant. Our studies send a fundamental message that, in using fluorescent reporters to explore changes in $[Ca^{2+}]_i$ during exposure to oxidants, toxins, and relatively uncharacterized neurohumoral agonists, it is important to take into account accumulation of interfering polyvalent cations.

In summary, we have adapted fluorometric approaches for monitoring changes in $[Ca^{2+}]_i$ in isolated glands of the rabbit gastric mucosa during exposure to oxidant stress. When contributions of interfering metal cations are controlled, it appears that NH₂Cl elicits increases in $[Ca^{2+}]_i$ that are sustained and not necessarily higher than those expected from normal signaling processes. These increases reflect contributions from emptying of physiologically regulated intracellular stores, as well as store-operated entry of extracellular Ca^{2+} . Our studies also indicate that these increases are mediated by oxidation of sulfhydryl (thiol) groups that are probably targeted by other endogenously generated oxidants, for example, nitric oxide. Further studies will determine the role played by Ca^{2+} signaling in the cellular response to thiol oxidants produced during acute tissue injury.

GRANTS

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